## ME 405 Fall 2006 Professor John M. Cimbala Lecture 18 10/18/2006

## Today, we will:

- Discuss Dilution Ventilation with Steady 100% Make-up Air in Section 5.3
- Do some example problems
- Discuss Dilution Ventilation with Unsteady Properties in Section 5.4
- Do some example problems



Solution: Since it is a standard it order ODE we can follow it  
. If A is B are construct 
$$\begin{bmatrix} Q, C_{e}, S & are construct \\ (\forall is always constant) \\ (\forall is always constant) \\ \vdots Andytical dold (fee Chap. 1) \\ \hline \\ \hline \\ C_{17} - C(s) = e^{-At} \\ \hline \\ C_{17} - C(o) \\ \end{bmatrix}$$
 it where  $\begin{bmatrix} C_{17} = B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} = B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} = B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} = B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} = B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} - C_{17} \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} - B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} - B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} - B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is where  $\begin{bmatrix} C_{17} - B \\ A \\ C_{17} - C(o) \\ \end{bmatrix}$  is the observation of the servation of the constant of the const

## **Example (Example 5.2 in text – When is it Safe to Enter a Room)**

**Given**: A fire has occurred in a hotel room with volume V = 85. m<sup>3</sup>. Immediately after the fire is extinguished, the mass concentration of hydrogen cyanide (HCN) is 10 g/m<sup>3</sup>. The firemen blow "fresh" air into the room at Q = 28.3 m<sup>3</sup>/min. Since there is some smoke outside the building, the "fresh" air actually has an ambient mass concentration of HCN equal  $c_a = 1.0$  mg/m<sup>3</sup>.

**To do**: Calculate how long the firemen need to wait before entering the room in order to stay below the STEL of HCN.

Solution: 
$$M_{JOJ} \rightarrow STEL = 5 \frac{mg}{m^3}$$
  
 $G_{Vra}: C(s) = 10 \frac{g}{m^3}$   
 $= 10,000 \frac{mg}{m^3}$   
 $= 10,000 \frac{mg}{m^3}$   
 $C_{a} = S = 0$  (fire is with)  
 $S_{Inc} = Q, S = 0$  (fire is with)  
 $C_{sj} = C_a + \frac{g}{Q} \rightarrow C_{sj} = C_a = 1.0 \frac{mg}{m^3}$   
 $eq: C_{sj} - C_{a} = exp(-\frac{Qt}{T})$   $\rightarrow Jabre for t$   
 $t = -\frac{V}{Q} \ln \left(\frac{C_{sj}-c}{C_{sj}-c(s)}\right) = 24 \frac{min}{s}$ 

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Next time - I will show examples using Runge-Kutta (R-K) to solve these kinds of unsteady problems